## A STEREOSPECIFIC SYNTHESIS OF <u>CIS</u>-B-LACTAMS<sup>1</sup> Ajay K. Bose, B. Dayal, H. P. S. Chawla, and M. S. Manhas Department of Chemistry and Chemical Engineering Stevens Institute of Technology Hoboken, New Jersey 07030

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A widely used method of  $\beta$ -lactam synthesis consists of the reaction of an acid chloride with an imine in presence of a base. The stereochemistry of the  $\beta$ -lactam formed in such a reaction is unpredictable – in most cases the product is either a <u>trans</u>- $\beta$ -lactam, or a mixture of <u>cis</u>- and <u>trans</u>- $\beta$ -lactams.<sup>2</sup> Other synthetic approaches, for example the reaction of an  $\alpha$ bromoester with a Schiff-base and zinc<sup>3</sup> or the photolysis of an unsaturated amide,<sup>4</sup> also give mixtures of isomers or only the <u>trans</u>-isomer. We wish to report a facile method for the stereospecific synthesis of <u>cis</u>- $\beta$ -lactams.

The reaction of the thioimidate (2) (prepared by treating the thioamide (1) with methyl iodide under alkaline conditions) with phenoxyacetyl chloride and triethylamine gave a single B-lactam (3)<sup>5</sup> in 80-90% yield.<sup>6</sup> Desulfurization of (3) with Raney nickel in boiling ethanol produced in about 50% yield a new  $\beta$ -lactam (4a) which was identified as <u>cis</u>-1,4-diphenyl-3-phenoxyazetidin-2-one described previously.<sup>7</sup> In a similar fashion several other <u>cis</u>- $\beta$ -lactams (4b, 4c, 4d) were prepared.

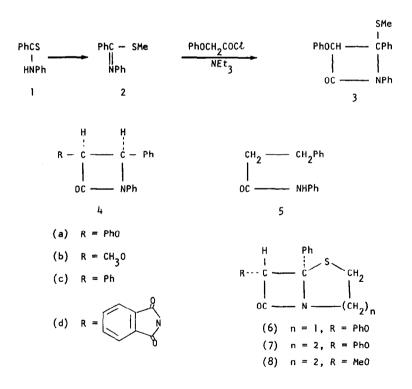
A secondary product in the desulfurization reaction was the amide (5) which could arise by the cleavage of the  $C_4$ -S,  $C_4$ -N and  $C_3$ -OPh bonds. The use of acetone in place of ethanol as the solvent made the desulfurization reaction milder, improved the yields and reduced the cleavage of the  $\beta$ -lactam ring.

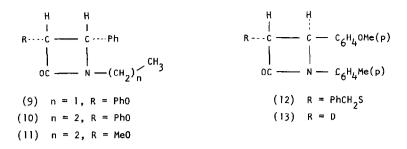
Using the appropriate cyclic imines, single isomers of fused  $\beta$ -lactams (6), (7), (8) incorporating a thiazolidine or a thiazine ring were obtained which upon desulfurization gave <u>cis</u>- $\beta$ -lactams (9), (10) and (11) respectively. The sulfoxides corresponding to (6), (7) and (8) also underwent easy desulfurization to afford cis- $\beta$ -lactams (9), (10) and (11).

In case of the cepham derivative (8), the resonance of the -0Me protons occurred at higher field ( $\delta$  3.10) than usual indicating that the phenyl group at the ring junction and the methoxy group were <u>cis</u> to each other. Since the cepham (8) and the corresponding sulfoxide

afforded a <u>cis</u>- $\beta$ -lactam, desulfurization must occur with retention of configuration.<sup>8</sup> The retention of configuration in these reactions was further confirmed by the desulfurization of the <u>trans</u>- $\beta$ -lactam (12) using deuterated Raney nickel in CH<sub>3</sub>OD. The product (13) showed about 70% deuterium incorporation as determined by mass spectroscopy and retained its <u>trans</u> stereochemistry. It can be concluded therefore that all the  $\beta$ -lactams prepared from thioimidates by treatment with acid chlorides have E-configuration. On the basis of the current views regarding the mechanism of  $\beta$ -lactam formation<sup>9</sup> it is not easy to account for the directive influence of the thioether function resulting in stereospecificity in  $\beta$ -lactam formation.<sup>10</sup>

The  $\beta$ -lactams in penicillin and cephalosporin C are <u>cis</u> in stereochemistry; their <u>trans</u>isomers have little biological activity.<sup>11</sup> The preparative method described here for <u>cis</u>- $\beta$ lactams is therefore of potential value to synthetic and medicinal chemistry – especially as naturally occurring monocyclic  $\beta$ -lactams have been discovered recently.<sup>12</sup>





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